

## X-ray Structural Analysis of Hexakis[2-(3,5-di-*t*-butylphenyl)-2-methylpropyl]diplobane

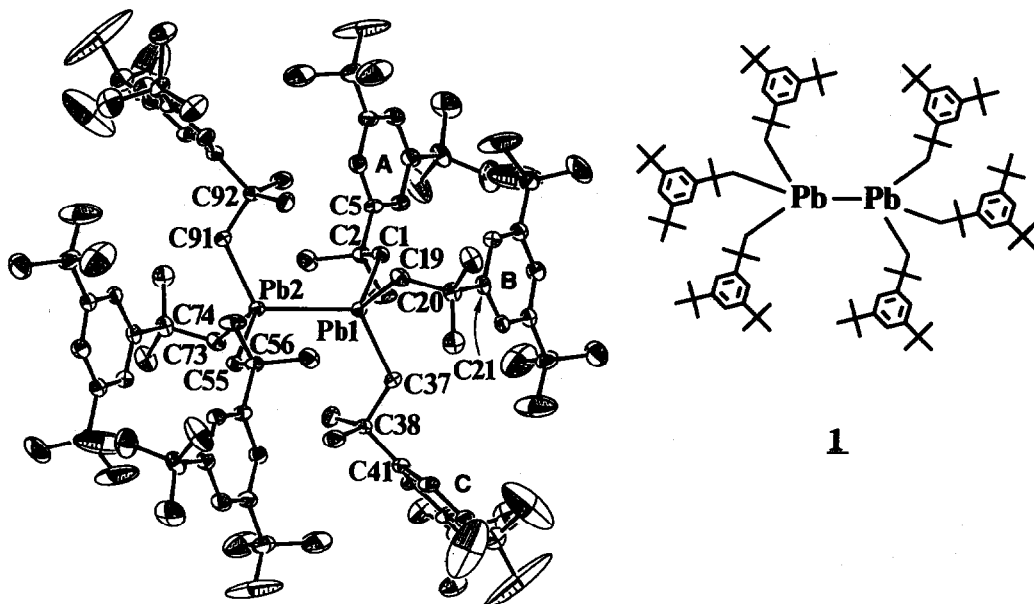
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**Abstract:** X-ray crystallographic analysis was performed for diplobane  $\text{Ar}'_6\text{Pb}_2$  ( $\text{Ar}'=2-(3,5\text{-di-}t\text{-butylphenyl})\text{-2-methylpropyl}$ ), prepared from the reaction of  $\text{ArM}$  ( $\text{Ar}=2,4,6\text{-tri-}t\text{-butylphenyl}$ ,  $\text{M}=\text{Li}$  or  $\text{MgBr}$ ) with  $\text{PbCl}_2$ , to show that it has the longest Pb-Pb distance ( $\text{Pb1-Pb2}=2.9448(6)$  Å) ever reported.

We have previously found that reactions of  $\text{ArM}$  ( $\text{Ar}=2,4,6\text{-tri-}t\text{-butylphenyl}$ ,  $\text{M}=\text{Li}$  or  $\text{MgBr}$ ) with  $\text{PbCl}_2$  afforded unusual products which contained rearranged  $\text{Ar}'$  group ( $\text{Ar}'=2-(3,5\text{-di-}t\text{-butylphenyl})\text{-2-methylpropyl}$ ).<sup>1</sup> Diplobane  $\text{Ar}'_6\text{Pb}_2$  (**1**), one of unexpected rearranged products, is especially interesting from a structural point of view. We report here the X-ray structure analysis for **1**.

The chemistry of group 14 ethane congeners  $\text{R}_3\text{X-YR}_3$  ( $\text{X}, \text{Y}=\text{C}, \text{Si}, \text{Ge}, \text{Sn}, \text{or Pb}$ ) have been of great interest for long time. It involves structural discussions for sterically congested compounds such as phantom hexaphenylethane (HPE)<sup>2</sup> which actually exists as the methylenecyclohexadiene structure.<sup>3</sup> The lead



**Figure 1.** ORTEP drawing of  $\text{Ar}'_6\text{Pb}_2$  (**1**). Selected bond lengths (Å) and angles (deg);  $\text{Pb1-Pb2}$  2.9448(6),  $\text{Pb1-C1}$  2.289(7),  $\text{Pb1-C19}$  2.280(7),  $\text{Pb1-C37}$  2.274(7),  $\text{Pb2-Pb1-C1}$  115.5(2),  $\text{Pb2-Pb1-C19}$  106.4,  $\text{Pb2-Pb1-C37}$  118.4(2),  $\text{C1-Pb1-C19}$  97.1(3),  $\text{C19-Pb1-C37}$  107.0(3),  $\text{Pb1-C1-C2}$  123.1(5),  $\text{Pb1-C19-C20}$  120.4(5),  $\text{Pb1-C37-C38}$  123.1(5). Selected torsion angles (deg);  $\text{Pb1-Pb2-C55-C56}$  63.1(5),  $\text{Pb1-Pb2-C73-C74}$  -150.7(5),  $\text{Pb1-Pb2-C91-C92}$  -11.3(7),  $\text{Pb1-C1-C2-C5}$  165.8(4),  $\text{Pb1-C19-C20-C21}$  173.4(5),  $\text{Pb1-C37-C38-C41}$  -177.1(5),  $\text{C1-Pb1-Pb2-C73}$  73.5(3),  $\text{C1-Pb1-Pb2-C91}$  -51.1(3),  $\text{C19-Pb1-Pb2-C91}$  55.4(3). Dihedral angles between least squares planes (deg); A-B 1.03, A-C 121.61, B-C 121.43.

analogue of HPE, hexaphenylplumbane  $\text{Ph}_3\text{PbPbPh}_3$  has been also investigated.<sup>4,5</sup> The discussions have focused on the structure in solution and in solid. Equilibriums such as  $(\text{R}_3\text{Pb})_2 \rightleftharpoons 2 \text{R}_3\text{Pb} \cdot$ <sup>4a,c</sup> or  $(\text{R}_3\text{Pb})_2 \rightleftharpoons \text{R}_4\text{Pb} + \text{R}_2\text{Pb}$ <sup>6</sup> have been postulated, but it was proven that  $\text{R}_3\text{PbPbR}_3$  was the appropriate structure for the description of the molecule.<sup>7</sup>

We became interested in the structure of the sterically crowded diplumbane  $\text{Ar}'_6\text{Pb}_2$ , and X-ray crystallographic analysis was performed (Figure 1).<sup>8</sup>  $\text{Ar}'_6\text{Pb}_2$  (**1**) has a very long Pb-Pb distance (2.9448(6) Å) and this is the longest lead-lead single bond so far reported for diplumbanes;  $\text{Me}_6\text{Pb}_2$  2.88(3) Å (electron diffraction<sup>10</sup>),  $\text{Ph}_6\text{Pb}_2$  2.848(4) Å (X-ray<sup>4d</sup>), (p-tolyl)<sub>6</sub>Pb<sub>2</sub> 2.851(1) Å (X-ray<sup>11</sup>), (cyclohexyl)<sub>6</sub>Pb<sub>2</sub> 2.876(4) Å (X-ray<sup>12</sup>), (o-tolyl)<sub>6</sub>Pb<sub>2</sub> 2.895(2) Å (X-ray<sup>9</sup>).

The Pb-Pb distance is apparently lengthened by the severe steric repulsion of Ar' groups on each lead atom. Three Ar' groups are not equivalent in the solid state, and the ring C lies nearly perpendicular to the rings A and B and in the space between the parallel rings A and B (see the plane-plane angle in Figure 1). Sebald and Harris<sup>9</sup> reported the <sup>207</sup>Pb CP MAS NMR of several sterically crowded diplumbanes, but among them the single crystal of (Mes<sub>3</sub>Pb)<sub>2</sub> (Mes=mesityl), whose X-ray crystallographic data would be necessary for comparison, has been unavailable because of its limited solubility.  $\text{Ar}'_6\text{Pb}_2$  would be useful to such CP MAS studies.

### References and Notes

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8. Crystallographic data for **1**:  $\text{C}_{108}\text{H}_{174}\text{Pb}_2$ ,  $M=1886.96$ , monoclinic,  $a=10.250(2)$ ,  $b=28.194(3)$ ,  $c=17.870(2)$  Å,  $\beta=94.66(1)^\circ$ ,  $V=5147(1)$  Å<sup>3</sup>,  $Z=2$ , space group  $P2_1/n$ ,  $D_c=1.217$  gcm<sup>-3</sup>,  $\mu=33.32$  cm<sup>-1</sup>. The intensity data ( $2\theta \leq 60^\circ$ ) were collected on a Rigaku AFC5R diffractometer with graphite monochromated Mo-K $\alpha$  radiation ( $\lambda=0.71069$  Å) and the structure was solved by direct methods. The non-hydrogen atoms were refined anisotropically. The final cycle of full-matrix least-squares refinement was based on 7621 observed reflections [ $I > 3.0\sigma(I)$ ] and 496 variable parameters with  $R(R_w)=0.047(0.046)$ . Atomic coordinates, bond length, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre.
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